

Michael Bolte<sup>a\*</sup> and  
Peyman Sakhaii<sup>b</sup><sup>a</sup>Institut für Anorganische Chemie, J. W. Goethe-Universität Frankfurt, Marie-Curie-Straße 11, 60439 Frankfurt/Main, Germany, and <sup>b</sup>Institut für Organische Chemie, J. W. Goethe-Universität Frankfurt, Marie-Curie-Straße 11, 60439 Frankfurt/Main, GermanyCorrespondence e-mail:  
bolte@chemie.uni-frankfurt.de

## Key indicators

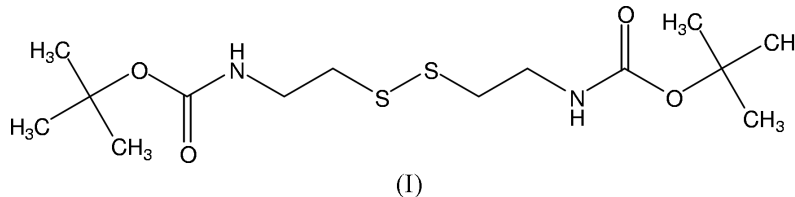
Single-crystal X-ray study  
 $T = 173$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å  
 $R$  factor = 0.046  
 $wR$  factor = 0.093  
Data-to-parameter ratio = 20.5For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.*N,N'*-bis(*tert*-butoxycarbonyl)cystamineThe title compound, Di-Boc-cystamine,  $\text{C}_{14}\text{H}_{28}\text{N}_2\text{O}_4\text{S}_2$ , serves as a building block for the synthesis of peptides. The molecule is located on a twofold rotation axis running through the centre of the S—S bond. The crystal packing is stabilized by N—H··O hydrogen bonds.

Received 24 August 2004

Accepted 31 August 2004

Online 4 September 2004

## Comment

The title compound, (I), serves as a building block for the synthesis of thioethyl-modified peptides (Moree *et al.*, 1993). The molecule (Fig. 1) possesses  $C_2$  symmetry, with a twofold rotation axis running through the centre of the S—S bond. Bond lengths and angles can be regarded as normal (Cambridge Structural Database, Version 1.6 plus three updates; *MOGUL* Version 1.0; Allen, 2002). The crystal packing (Fig. 2) shows a ladder-like structure in which adjacent molecules are held together by N—H··O hydrogen bonds.

## Experimental

The title compound, (I), was synthesized according to the procedure described by Moree *et al.* (1993). Colourless crystals were grown from an ethanol solution.

## Crystal data

 $\text{C}_{14}\text{H}_{28}\text{N}_2\text{O}_4\text{S}_2$   
 $M_r = 352.50$   
Monoclinic,  $C2/c$   
 $a = 19.674$  (2) Å  
 $b = 9.6968$  (8) Å  
 $c = 9.8462$  (9) Å  
 $\beta = 92.589$  (5)°  
 $V = 1876.5$  (3) Å<sup>3</sup>  
 $Z = 4$  $D_x = 1.248$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
Cell parameters from 510 reflections  
 $\theta = 3.1$ – $19.7^\circ$   
 $\mu = 0.30$  mm<sup>-1</sup>  
 $T = 173$  (2) K  
Needle, colourless  
 $0.45 \times 0.06 \times 0.04$  mm

## Data collection

Siemens SMART CCD three-circle diffractometer  
 $\omega$  scans  
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  
 $T_{\min} = 0.877$ ,  $T_{\max} = 0.988$   
7862 measured reflections2127 independent reflections  
1271 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.067$   
 $\theta_{\text{max}} = 28.3^\circ$   
 $h = -26 \rightarrow 24$   
 $k = -12 \rightarrow 12$   
 $l = -13 \rightarrow 12$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.046$   
 $wR(F^2) = 0.093$   
 $S = 1.02$   
 2127 reflections  
 104 parameters  
 H atoms treated by a mixture of  
 independent and constrained  
 refinement

$$w = 1/[\sigma^2(F_o^2) + (0.0327P)^2 + 0.5998P]$$

where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.21 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.27 \text{ e } \text{\AA}^{-3}$

Table 1

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

S1—C2	1.809 (2)	C5—O51	1.220 (2)
S1—S1 <sup>i</sup>	2.0448 (11)	C5—O6	1.348 (2)
C3—N4	1.449 (3)	O6—C7	1.475 (2)
N4—C5	1.343 (3)		
C2—S1—S1 <sup>i</sup>	103.12 (7)	O51—C5—O6	125.6 (2)
C3—C2—S1	114.38 (15)	N4—C5—O6	110.54 (18)
C5—N4—C3	120.93 (19)	C5—O6—C7	120.56 (16)
O51—C5—N4	123.9 (2)		
C2 <sup>i</sup> —S1 <sup>i</sup> —S1—C2	65.13 (16)		

Symmetry code: (i)  $1 - x, y, \frac{1}{2} - z$ .

Table 2

Hydrogen-bonding geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N4—H4 $\cdots$ O51 <sup>ii</sup>	0.79 (2)	2.14 (2)	2.866 (2)	152 (2)

Symmetry codes: (ii)  $x, 1 - y, \frac{1}{2} + z$ .

H atoms bonded to C atoms were refined with fixed individual displacement parameters [ $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}_{\text{methylene}})$  or  $1.5U_{\text{eq}}(\text{C}_{\text{methyl}})$ ], using a riding model, with  $C-H = 0.99$  and  $0.98 \text{ \AA}$ , for methylene and methyl H atoms, respectively. The H atom bonded to nitrogen was refined isotropically.

Data collection: *SMART* (Siemens, 1995); cell refinement: *SMART*; data reduction: *SAINT* (Siemens, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* in *SHELXTL-Plus* (Sheldrick, 1991); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2003).

## References

- Allen, F. H. (2002). *Acta Cryst.* **B58**, 380–388.  
 Moree, W. J., van Gent, L. C., van der Marel, G. A. & Liskamp, R. M. J. (1993). *Tetrahedron*, **49**, 1133–1150.  
 Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.  
 Sheldrick, G. M. (1991). *SHELXTL-Plus*. Release 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.  
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.

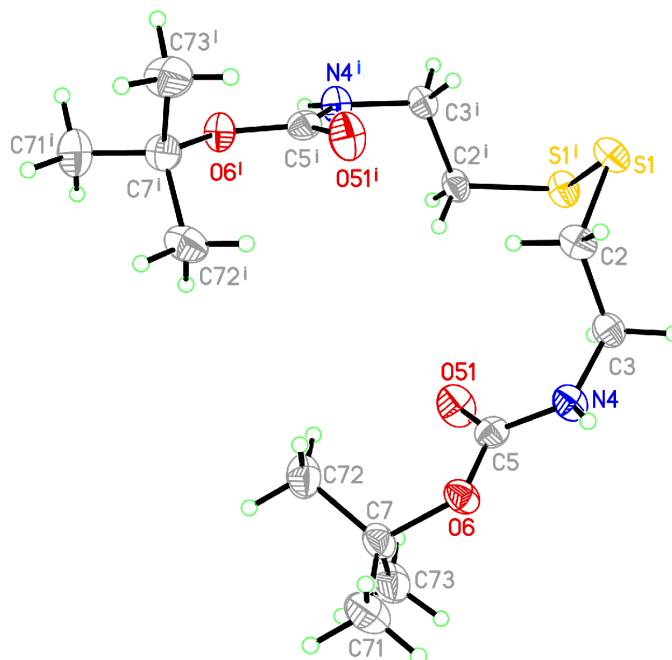


Figure 1

Perspective view of (I), with the atom numbering. Displacement ellipsoids are drawn at the 50% probability level. Symmetry code (i) as in Table 1.

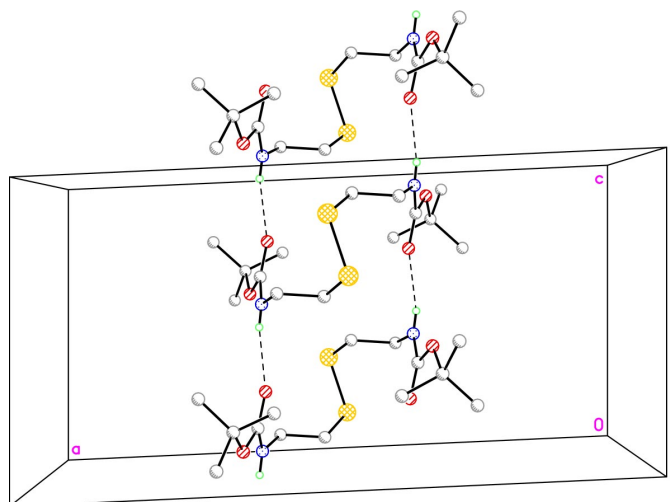


Figure 2

Packing diagram of (I), viewed on the  $ac$  plane. Hydrogen bonds are shown as dashed lines.

- Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.  
 Siemens (1995). *SMART* and *SAINT*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.  
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.